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Alumina–aluminum interpenetrating-phase composites with three-dimensional periodic architecture

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Abstract

Robotic deposition was used to create an alumina structure with three-dimensional periodicity and submillimeter feature size. Liquid metal infiltration of this structure resulted in an Al_2O_3 -Al interpenetrating-phase composite exhibiting low thermal expansion and high compressive strength.

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1. Introduction

Interpenetrating phase composites (IPCs) represent a family of materials whose microstructure is characterized by the continuity and interpenetration of two or more phases. In the context of ceramic–metal IPCs, much of the driving force for investigating interpenetrating microstructures has been the toughening of ceramics by addition of low concentrations of a metal phase; therefore, IPCs of 60 vol.% ceramic or more are most common in the literature. The Al₂O₃–Al system is one of the most studied ceramic–metal IPC systems. Several processing strategies have been developed to produce these composites, including infiltration of porous Al_2O_3 preforms [1–3], reactive metal penetration [4] and displacement reactions [5,6]. Other processing routes may result in discontinuous or partially discontinuous phases (e.g., directed metal oxidation [7] and powder metallurgy of mixed powders). All Al_2O_3 –Al IPCs produced by these processes have a random, usually isotropic, spatial distribution of phases.

New opportunities for creating ceramic-metal IPCs with a highly regular architecture and tailored properties exist through the recent development of solid freeform fabrication techniques. Complex three-dimensional (3-D) ceramic architectures can be fabricated in a layerwise manner by fused deposition [8,9] and direct-write methods [10–15]. Direct-write techniques, such as robocasting [11–13], ink-jet printing [14], and micro-pen writing

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[15], involve layer-by-layer assembly of colloidal inks. The fabrication of 3-D periodic structures with spanning (unsupported) features place the most stringent demands on ink design [12,13]. In the present paper, we report on an interpenetrating Al_2O_3 -Al composite produced by liquid metal infiltration of a 3-D periodic Al_2O_3 preform with spanning elements that was produced by robotic deposition.

2. Experimental procedures

2.1. Alumina tower fabrication

Al₂O₃ powder (AKP-15, Sumitomo Chemical Co., NY, with a specific surface area of $3.9 \text{ m}^2/\text{g}$, density of 3.97 g/cm³, and mean particle size of 0.8 μ m), and ZrO₂ powder (3Y-TZ, Tosoh Corp, Tokyo, Japan, with a specific surface area of 14.3 m^2/g , density of 5.89 g/cm³, and mean particle size of 0.4 μ m) were mixed with a volume ratio of 95:5. A 30 vol.% ceramic suspension was created by adding the mixed powders to a 40% aqueous solution of an ammonium polyacrylate polymer (Darvan 821A, from Vanderbilt Co, Norwalk, CT). This stable suspension was vigorously agitated for 1 h to ensure homogeneity and then centrifuged at 1500 rpm for 1 h. After centrifugation, the supernatant was removed to increase the solids loading to approximately 54 vol.%. Methvlcellulose (Methocel F4M, from Dow Chemical, Midland, MI) was added (7 mg/ml of ceramic) as a viscosifying aid. Gelation was induced by first adding 1.5 M AlCl₃ and then lowering the pH via addition of 1 M HNO₃. The pH was adjusted to roughly 8.5 with 1 M HNO₃ and 1.5 M NH₄OH to achieve the desired equilibrium shear modulus G'of ~ 1 MPa. The colloidal gel-based ink contained approximately 52 vol.% solids.

Two Al₂O₃ towers, consisting of 30 alternating 0/90 layers of parallel rods, were produced by robotic deposition using the gel-based ink described above. The ink delivery system was mounted on a *z*-axis motion control stage that prints onto a moving x-y stage. The entire process was controlled through a custom-designed, computer-aided, direct-write program (RoboCad 3.2). The deposition

nozzle (330 μ m in diameter) was coated with nonwetting oil, which deposits onto the extruded ink filament to inhibit drying during the build sequence. After assembly, the 30-layer tower structures were dried in air for 24 h and then sintered in air at 1600 °C for 2.5 h. The final rod diameter was approximately 250 μ m in the densified structures.

Fig. 1a is a top view of a sintered Al₂O₃ tower with nominal dimensions of $6 \times 4 \times 4$ mm³. Fig. 1b is a schematic illustrating how the tower was created. The first layer is composed of a series of equidistant rods parallel to the *x*-axis with "hairpins" connecting them. The second layer is identical to the first but rotated by 90°, i.e., parallel to the *y*-axis. Use of gel-based inks allows the deposited filamentary rods to maintain their shape without deformation as they span gaps between rods in underlying layers [12,13]. This process was repeated 15 times, resulting in a 30-layer tower structure with simple cubic symmetry. Fig. 2a and b are a side view and 3-D schematic of the tower



Fig. 1. (a) Top view of an alumina tower and (b) an idealized schematic illustrating the layering pattern.



Fig. 2. (a) Side view of the tower architecture and (b) an idealized 3-D schematic.

architecture, respectively. The towers consist of vertical columns (parallel to the *z*-axis) and horizontal spans (parallel to the *x*- and *y*-axes). Vertical columns are formed at the interface between horizontal rods in one layer and those in the adjacent layers, as they partially fuse together during deposition [12]. The unsupported regions of the deposited rods form spans that connect columns together within the 3-D periodic Al_2O_3 tower.

2.2. Composite processing

Sintered Al₂O₃ towers were placed within cavities machined into a graphite block; the dimensions of each cavity were slightly larger than an individual tower (by about 1 mm on a side). The graphite block was placed within an alumina crucible and a rod of 99.99% pure aluminum was placed on top of the graphite. The crucible was put in a gas-pressure, liquid-metal infiltration apparatus [16] and heated under vacuum to a temperature of 750 °C. The liquid metal was infiltrated into the evacuated open volume of the tower under an argon pressure of 3.5 MPa and the resulting composite was directionally solidified. A skin of pure aluminum, nominally 0.5 mm in thickness, encapsulated the composite and was removed from the two $4 \times 4 \text{ mm}^2$ sides by light mechanical polishing to ensure uniform loading during subsequent mechanical testing. This skin was retained on the other four $(4 \times 6 \text{ mm}^2)$ sides to prevent mechanically damaging the Al₂O₃ architecture.

2.3. Property measurements

A calibrated horizontal push-rod dilatometer (Orton Model 1600, Westerville OH) was used to measure the thermal expansion of an Al₂O₃ tower and the Al₂O₃–Al composite in the z-direction by heating to 200 °C at a rate of 3 °C/min. After the dilatometric experiments, the compressive strength of both tower and composite was measured at constant crosshead speed, corresponding to a nominal strain rate of 10^{-3} s⁻¹. Both the Al₂O₃ tower and the Al₂O₃–Al composite were loaded in the z-direction, so that the columns created at the contact points between layers were parallel to the loading direction. The compression platens were lubricated with Teflon and two methods were used to determine the strain in the sample: laser extensometry and the crosshead motion corrected for the compliance of the testing machine. Neither method produced data with sufficient resolution for stiffness determination.

3. Results and discussion

The density of the fired elements in the Al_2O_3 towers was determined by helium pycnometry to be 3.7 g/cm³, which was lower than expected for a dense mixture of Al₂O₃-5 vol.% ZrO₂ (4.07 g/cm³ estimated from rule of mixtures). This value corresponds to a closed porosity of as much as 9% within the ceramic that cannot be infiltrated (subsequently fabricated towers exhibit densities greater than 95% of theoretical). The overall density of the towers was estimated from their mass and dimensions as 2.6 g/cm³; the volume fraction of ceramic in the tower is then calculated as 70%. The density of the Al₂O₃-Al composite was measured as 3.4 g/cm³ by helium pycnometry (this value is corrected for the Al skin, by subtracting its contribution based on the dimensions of the composite compared to those of the tower before infiltration). Using the density of Al (2.69 g/cm^3) , the ceramic (3.7 g/cm^3) and the composite (3.4 g/) cm^3), the concentration of Al₂O₃ in the composite is calculated to be 70 vol.%. The good agreement with the previous value of 70 vol.% indicates that infiltration was complete and that all open porosity in the tower was filled with Al.

3.1. Thermal expansion

The average coefficient of thermal expansion (CTE) of the as-sintered Al₂O₃ tower between 50 and 200 °C is $6.7 \pm 0.7 \times 10^{-6}$ K⁻¹, as measured by dilatometry (near room temperature, the CTE of the tower is about 6×10^{-6} K⁻¹). This value is consistent with CTE reported for bulk Al₂O₃ in this temperature range (5–7×10⁻⁶ K⁻¹ [1,17]). The CTE of the Al₂O₃–Al composite was measured to be $8.9 \pm 0.3 \times 10^{-6}$ K⁻¹ between 50 and 200 °C.

Plasticity during thermal expansion, which tends to increase the CTE value as compared to

Table 1

elastic predictions [18], is not relevant here, because the relatively small temperature excursion $(\Delta T = 175 \text{ °C})$ results in a representative mismatch strain $\Delta \alpha \Delta T$ of only about 0.4%. Assuming that, upon cooling from the processing temperature, the matrix had reached its yield stress in tension and was work-hardened, this small mismatch strain during CTE measurement is unlikely to induce significant plasticity, so that modeling the elastic interaction of the constituents should provide reasonable predictions for thermal expansion.

Analytical solutions for the CTE of isotropic discontinuously reinforced composites (DRCs) have been developed by considering the thermal elastic interactions of an inclusion surrounded by matrix in specific volumetric proportions (see review in Ref. [19]). Two limits can be identified depending on which phase is the inclusion. These bounds are often referred to as the Kerner (upper) bound and the Schapery (lower) bound. The upper bound for the CTE represents the metal matrix composite configuration, i.e., in the context of this study, Al₂O₃ particles in a continuous Al matrix. The ceramic matrix configuration, with Al inclusions in an Al₂O₃ matrix, is captured by the lower bound, where the continuous Al₂O₃ matrix significantly constrains the expansion of the discontinuous metal.

Although these inclusion models do not reflect the interpenetrating microstructure of an IPC, a solution for the CTE of an interpenetrating microstructure can be obtained by self-consistently solving the equations for the two bounding geometries [20], the so-called classical self-consistent (CSC) approach. This solution, as well as numerical solutions [21,22], fall within the elastic bounds and are typically close to the lower elastic bound (i.e., near the solution for a ceramic matrix composite). Using the material properties listed in Table 1, the upper and lower elastic bounds are calculated to be 10.8 and 9.2×10^{-6} K⁻¹ respectively. The measured CTE $(8.9 \times 10^{-6} \text{ K}^{-1})$ is, within experimental error, equal to the lower elastic limit, as anticipated since the alumina is continuous. Other researchers have reported the CTE of IPCs with 70 ± 5 vol.% Al₂O₃ to be closer to the upper bound, around 10×10^{-6} K⁻¹ [1,5,6]; exact comparison is

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Prope	erties of	pure	aluminum	and	alumina,	average	between	50
and 2	200 °C							

	Al	Al_2O_3
Young's modulus, <i>E</i> (GPa)	63.5ª	410 [17]
Poisson's ratio (v–)	0.35	0.235 ^a
Shear modulus, G (GPa)	23.5 [28]	166 [17]
Bulk modulus, K (GPa)	70.6 ^a	258 ^a
CTE ($\alpha \times 10^{-6} \text{ K}^{-1}$)	24.8 [29]	6.2 [17]

^a Determined from elastic relationships: $G = E/(2+2\nu)$; $K = E/(3-6\nu)$.

not possible since the CTE was not reported in the same temperature range.

Metal-ceramic composites are used for thermal management applications, such as electronic packaging, where CTE values close to silicon $(4.1 \times 10^{-6} \text{ K}^{-1})$ and alumina $(6.7 \times 10^{-6} \text{ K}^{-1})$ are desired, coupled with thermal conductivities approaching that of aluminum alloys [23]. Although the thermal conductivity of the present infiltrated tower was not measured, the use of high purity aluminum, the coarse size of the constituents and the spatial periodicity of the composite (the Al "columns" in the composite act as "heat pipes"), all point towards improved thermal conductivity as compared to IPCs described in the literature, which typically exhibit an alloyed aluminum, smaller architectural size scales and a random distribution of the constituents.

3.2. Mechanical properties

The compressive strength of the Al₂O₃ tower was measured as 190 MPa, which is low compared to typical values for dense Al₂O₃ (2.5–3.8 GPa), but similar to tensile and flexure strengths for bulk Al₂O₃ (250 and 400 MPa respectively [17]). Fracture stresses measured in four-point bending have been reported in the range 140 and 200 MPa for porous Al₂O₃ preforms with 70 ± 5 vol.% Al₂O₃ prepared for subsequent infiltration to produce IPCs [2,3]. Periodic alumina and mullite structures (70 vol.%) produced by fused deposition, however, have compression strengths that are less than 50 MPa [9,24]. Finite-element modeling of those periodic mullite structures shows that the continuous columnar elements support the applied compressive loads [25] while the spans are in tension, as they prevent buckling of the columns. Since alumina is nearly an order of magnitude stronger in compression than tension, failure is expected in the spans loaded in tension, resulting in spallation of layers parallel to the loading direction, as reported for the periodic mullite structures [25]. The same failure mechanism was observed in the present investigation, indicating that the majority of the microstructural damage was indeed generated in the horizontal spans. Sharp corners at the junctions between columns and spans are likely to initiate this failure at relatively low tensile stresses; failure stresses may also be lowered by porosity in the ceramic. This interpretation implies that failure of the tower is determined by the defects within the spans, not by the columns or the span lengths. We note that standard ceramic foam models [26] assume that all spans have a large length-to-thickness ratio, so that bending failure predominates; this assumption is not appropriate for the present towers.

The compressive strength (failure stress) of the composite was 700 MPa, an improvement by a factor of almost 4 over the un-infiltrated tower. It is unlikely that this improvement is due to a fourfold decrease in the load carried by the ceramic phase through load transfer to aluminum, given its much lower strength and stiffness as compared to alumina. Rather, the main effect of the metallic phase may be similar to that of the horizontal alumina spans: prevention of buckling in the load-bearing alumina columns (similarly, in unidirectional fiber composites tested in compression, all of the buckling resistance is supplied by the matrix). Metallographic observation of the deformed composite revealed that damage developed in a similar manner as in the unreinforced tower: failure occurred predominantly in the spans near the corners where the spans join the columns, Fig. 3a. As the spans fractured, the aluminum phase not only carries an increasing proportion of the transverse tensile stresses but also blunts cracks by plastic deformation, preventing catastrophic propagation of cracks within the alumina phase. On a macroscopic level, damage in the composite was found to link into a macroscopic



crack forming at an angle of about 45° with respect to the applied load, i.e., on the plane of maximal shear stress, as typically observed for brittle materials in compression, Fig. 3b.

Higher compressive strengths have been reported for Al_2O_3 -Al IPCs produced by displacement reactions [5] (up to 1200 MPa for 64 vol.% Al_2O_3 [6]). Numerous authors have reported bending strengths between 400 and 700 MPa for 65–75 vol.% Al_2O_3 -Al IPCs [2–5], implying compressive strengths of 700 MPa or greater. Those materials, however, have a much finer microstructure corresponding to smaller flaw size, and thus higher strength in the ceramic. The strength of sintered alumina is also dependent on



the sintering conditions and additives, making direct comparison difficult without detailed analysis of the sintered microstructure. The aluminum matrix, in the case of material produced by displacement reactions, has significant (and difficult to control) amounts of silicon, compared to the high-purity aluminum used in this study, again appreciably affecting mechanical and thermal properties. Mechanical data for mullite-aluminum IPCs produced by a process analogous to the one used here (fused deposition of the mullite followed by infiltration with aluminum [27]) have not been reported, although as mentioned above, the un-infiltrated mullite preforms displayed a compressive strength much lower than in the present study.

4. Summary

An interpenetrating-phase composite with a 3-D periodic architecture was created by pressure infiltration of an alumina tower with liquid aluminum. The alumina tower with periodic cubic symmetry was produced by robotic deposition of an Al₂O₃-5 vol.% ZrO₂ gel. The composite, consisting of 70 vol.% ceramic, exhibits a low thermal expansion of 8.9×10^{-6} K⁻¹, close to the lower elastic (Schapery) bound for thermal expansion. The compressive strength of the composite is much greater than that of the un-infiltrated ceramic preform (700 versus 190 MPa), with only a small increase in density (3.4 versus 2.6 g/cm³).

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